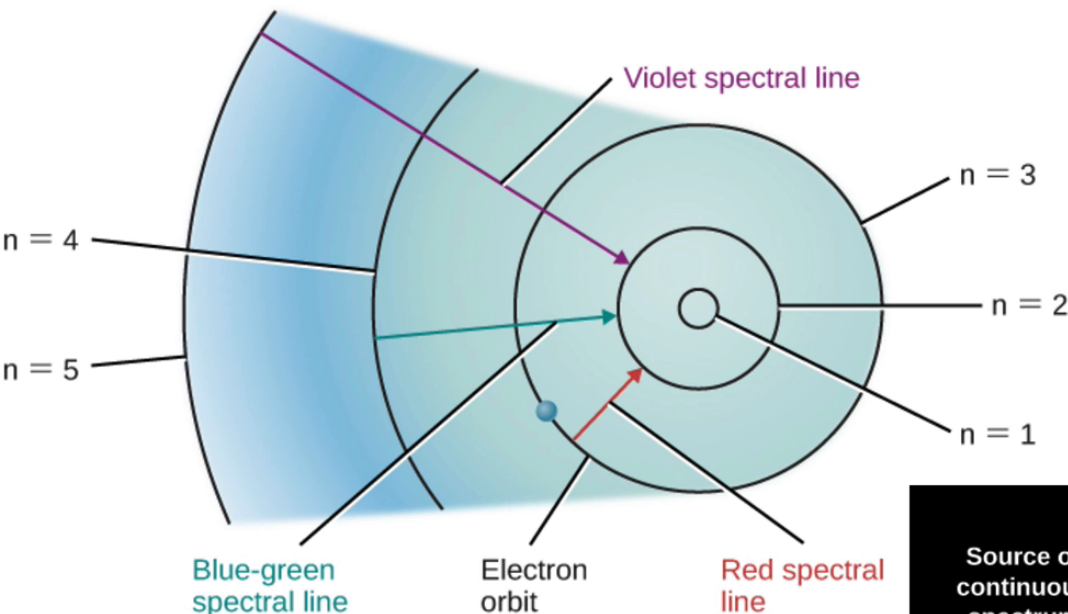


1 Bohr Model - Energy Levels Light and Atoms



Jump down to lower Energy
But Energy Conserved
Released as light

2 Light having same Energy can be absorbed or emitted

3 Same wavelength can be absorbed

“Excite” the atom

Emitted later on in any direction

Use Spectra to see gas clouds at specific wavelength/frequency

